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conseq number of moles of CaCO3 and MgCO3 = 0.01 conseq number of moles of CaCO3 = 0.005 [3] (ii) Calculate the volume of hydrochloric acid, 1.0 mole/dm3, needed to react with one tablet. number of moles of CaCO3 and MgCO3 in one tablet = 0.01 Expect same as answer to (c)(i). NO marks to be awarded. Just mark consequentially to this response

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