

## Bohr Model Introduction Gizmo Answers

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### Bohr Model Introduction Gizmo Answers

The Bohr Model: Introduction Gizmo™ explores this connection. The laser shown in the Gizmo can emit photons , or particles of light, at a variety of wavelengths. The energy of a photon, measured in electron volts (eV), is inversely proportional to its wavelength.

### Student Exploration Bohr Model- Introduction (ANSWER KEY)

The Bohr Model: Introduction Gizmo™ explores this connection. The laser shown in the Gizmo can emit photons, or particles of light, at a variety of wavelengths.

### Student Exploration Bohr Model- Introduction (ANSWER KEY ...

Bohr Model: Introduction Answer Key Vocabulary: absorption spectrum, Bohr model, electron volt, energy level, laser, orbital, photon Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

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Fire photons to determine the spectrum of a gas. Observe how an absorbed photon changes the orbit of an electron and how a photon is emitted from an excited electron. Calculate the energies of absorbed and emitted photons based on energy level diagrams. The light energy produced by the laser can be modulated, and a lamp can be used to view the entire absorption spectrum at once.

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Introduction: A catalyst is a ... Bohr Model of Hydrogen Vocabulary : absorption spectrum, Bohr model, electron volt, emission spectrum, energy level, ionization energy, laser, orbital, photon Prior Knowledge Questions (Answer these questions BEFORE using the Gizmo.) What happens to an electron when it absorbs a tiny packet of light, called a ...

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