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Data Analysis
Data
Analysis 1. A sample of
aluminum is placed in
a 25-mL graduated
cylinder containing
10.0 mL of water.

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Data
Analysis 1. A sample of
aluminum is placed in
a 25-mL graduated
cylinder containing
10.0 mL of water. The
level of water rises to
18.0 mL. Aluminum has
a density of 2.7 g/mL.
Calculate the mass of
the sample. 2.

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19 Gases
Gases 1. In
one city, a balloon with
a volume of 6.0 L is
filled with air at 101
kPa pressure. The
balloon is then taken
to a second city at a
much higher altitude.

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Data Analysis 1. A sample of aluminum is placed in a 25-mL graduated cylinder containing

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10.0 mL of water. The level of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL. Calculate the mass of the sample. 2. Saturn is about 1 429 000 km from the Sun.

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Structural Formulas of
Alkanes SG 22.1 &

22.2 Structures of

Alkenes & Alkynes SG

22.3 & 22.4 SG 23.1,

23.2, 23.3 Chapter 22

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Structures of

Functional Groups

Chapter 22 Review

Assessment-----

Chapter 3 SG 3.1 SG

3.2 SG 3.4 Chapter 3

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Changes ...

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weather balloon
contains 14.0 L of
helium at a pressure of
95.5 kPa and a
temperature of 12.0°C.
If this had been stored

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in a 1.50-L cylinder at 21.0°C, what must the pressure in the cylinder have been? 15. How many moles of a gas will occupy 2.50 L at STP? 16.

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PROBLEMS - Galax

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Data Analysis
Data Analysis 1. A sample of aluminum is placed in a 25-mL graduated cylinder containing 10.0 mL of water. The level of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL.
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AnalysisData Analysis

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a density of 2.7 ...

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Chapter 13 Chapter

Assessment Chapter 14

Boyle's and Charles' SG

14.1 & 14.4 Gay

Lussac's Mixed Review

Combined & Ideal

Partial Pressures SG

14.3 CHAPTER 3

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Supplemental Question
039 The internal shear
force V at a certain
section of an aluminum
beam is 10.7 kN.

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lectures a chapter 14,
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and a chapter 15,
'Mercury in the
Environment'. I have
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to support these

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chapters. All problems are from recent exams in my course. This 5 th edition includes a number of new problems and a few corrections to the previous (August 2011) edition.

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Answer Key Physics

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Chemistry: Matter and Change • Chapter 3 3
Matter—Properties and Changes
Properties and Changes
1. An 18-g sample of element A combines completely with a 4-g sample of element B to form the compound AB. What is the mass of the compound formed? 2.

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