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15.1: Classifications of Acids and Bases In chemistry, acids and bases have been defined differently by three sets of theories: One is the Arrhenius definition defined above, which revolves around the

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idea that acids are substances that ionize (break off) in an aqueous solution to produce hydrogen (H⁺) ions while bases produce hydroxide (OH⁻) ions in solution.

15: Acid-Base Equilibria - Chemistry LibreTexts

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Exercises: Oxtoby et al. - Chemistry LibreTexts

Chemistry Principles of Modern Chemistry In some reactions there is a competition between kinetic control and thermodynamic control over product yields. Suppose compound A can undergo two elementary reactions to stable products: $A \xrightleftharpoons[k_1]{k_2} B$ or $A \xrightleftharpoons[k_2]{k_1} C$ For simplicity we assume first-order kinetics for both forward and reverse reactions.

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