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### The Periodic Table and Periodic LawThe Periodic Table and ...

360 Chemistry: Matter and Change • Chapter 18 Solutions Manual CHAPTER 18 SOLUTIONS MANUAL Section 18.3 Hydrogen Ions and pH pages 650-658 Practice Problems pages 651-657 22. The concentration of either the H<sup>1</sup> ion or the OH<sup>2</sup> ion is given for four aqueous solutions at 298 K. For each solution, calculate [H<sup>1</sup>] or [OH<sup>2</sup>]. State whether the solution is acidic, basic, or neutral. a. [H<sup>1</sup>] 5 1.0 3 10 213 M K w 5 [H<sup>1</sup>][OH<sup>2</sup>]

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