

Unsaturated Hydrocarbons Study Guide

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Unsaturated Hydrocarbons Study Guide

Hydrocarbons are organic compounds made up of carbon and hydrogen atoms only. Hydrocarbons can be categorized as either saturated hydrocarbons or unsaturated hydrocarbons. Saturated hydrocarbons...

Unsaturated Hydrocarbon: Definition & Examples - Study.com

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Alkenes are hydrocarbons with a carbon-to-carbon double bond. 3.2: Bonding in Ethene; 3.3: Cis-Trans Isomers (Geometric Isomers) Cis-trans (geometric) isomerism exists when there is restricted rotation in a molecule and there are two nonidentical groups on each doubly bonded carbon atom. 3.4: Physical Properties of Alkenes

3: Unsaturated Hydrocarbons - Chemistry LibreTexts

Unsaturated hydrocarbons are extremely useful organic compounds in the manufacturing of plastics. LDPE, which is a low-density variation of polyethylene, is used in the manufacturing of grocery bags; Polystyrene is used in making egg cartons, disposable cups, and other convenient products.

Unsaturated Hydrocarbon - Definition, Examples, and Uses

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The definition of an alkene is an unsaturated hydrocarbon with one or more carbon-carbon double bonds within the molecule. Alkenes have the general formula: $C_n H_{2n}$. This means that in a simple alkene (only one double bond) there are twice as many hydrogen atoms as there are carbon atoms.

Alkenes and unsaturated

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The hydrogenation reaction is possible only for compounds with double or triple bonds, and such compounds are said to be unsaturated hydrocarbons. The addition of the hydrogen to the carbon atoms that were double- or triple-bonded converts the unsaturated compound to a saturated hydrocarbon with only single bonds.

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What is the term for a family of unsaturated hydrocarbon compounds having a triple bond? A) alkenes B) alkynes C) alkanes D) arenes E) none of the above. ... APS Exam Study Guide. 42 terms. Abby_Hayden3. Amendment Exam. 38 terms. Abby_Hayden3. Chem Chapter 9 Study Guide. 41 terms. Abby_Hayden3. Chem Chapter 2 Study Guide. 91 terms.

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Unsaturated hydrocarbons are hydrocarbons that have double or triple covalent bonds between adjacent carbon atoms. The term "unsaturated" means more hydrogen atoms may be added to the hydrocarbon to make it saturated (i.e. consisting all single bonds).

Unsaturated hydrocarbon - Wikipedia

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Study Guide Hydrocarbons study guide hydrocarbons Chapter 21: Hydrocarbons 744 Chapter 21 • Hydrocarbons Section 221111 Introduction to Hydrocarbons MAIN Idea Hydrocarbons are carbon-containing organic compounds that provide a source of energy and raw materials Real-World Reading Link If you have ridden in a car or a bus, you have used

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Chemists called the hydrocarbons that reacted with bromine unsaturated hydrocarbons in the same sense that an unsaturated aqueous solution can dissolve more solute. Hydrocarbons that did not react with bromine were said to be saturated. Present-day chemists can now explain the experimental results obtained 170 years ago.

Chapter 21: Hydrocarbons

Propene Question 21 2-Pentene Question 22 Ethene Question 23 Correct! 2-Butene Question 24 Unsaturated hydrocarbons: Contain at least one carbon-carbon multiple bond Question 25 Alkyne molecules a hydrocarbon that has one or more triple bonds. name

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ends in -yne. Question 26 Consider a straight chain hydrocarbon that has three carbon atoms, six hydrogen atoms, and no side chains.

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Assmt.6 Question1 YouAnswered ...

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Which compound is a saturated hydrocarbon? ethane, ethene, ethyne, ethanol. Saturated and Unsaturated Hydrocarbon The hydrocarbons that contain only single bonds are classified as saturated and the...

Which compound is a saturated hydrocarbon ... - study.com

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In all unsaturated hydrocarbons, all carbon atoms are sp hybridized. D In all saturated hydrocarbons A at least two of the carbon atoms are sp or sp hybridized D In all unsaturate D In all unsaturated hydrocarbons, at least two of the carbon atoms are either sp' or sp hybridized. d hydrocarbons, all carbon atoms are either sp or sp hybridized.

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